



Original Research Article

Evaluation of lugol's iodine as a screening tool for identification of dried saliva stains

Neetu Pandey^{1,*}, Rajendra Patil¹, Udit Singh¹, Himani Singh¹, Satarupa Debnath¹, Deeksha Gahlot¹¹Dept. of Dental, Kothiwal Dental College and Research Centre, Moradabad, Uttar Pradesh, India

ARTICLE INFO

Article history:

Received 28-01-2023

Accepted 04-02-2023

Available online 29-04-2023

Keywords:

Forensic science

Criminology

Body Fluids

Saliva

Lugol

s Iodine

ABSTRACT

Background: The evidential value of saliva in stains encountered in crime investigations can be of considerable importance. Whether the stains are a result of expectoration or a consequence of some sexual or violent act resulting in saliva being mixed with other body fluids, the investigating forensic expert is faced with two fundamental problems: localization of stain and visualization of stain. Visualization of saliva is problematic unlike other body fluids like blood which has its own internal marker (i.e., hemoglobin) or semen that can be easily identified by acid phosphatase method.

Aim: Evaluate of lugol's Iodine as an easy, chair side screening tool for identification of saliva from the crime scene.

Materials and Methods: In the present study saliva stains on different fabrics viz. silk, polyester, linen, jute, cotton, georgette, sarton and other items like paper cups, chocolate wrapper aluminium, plastic bags, stainless steel spoon (cutlery) were examined using Lugol's iodine.

Result: Lugol's Iodine test was found to give Positive results for saliva stains on different fabrics and it was also found that the stains on different fabrics gives positive results for saliva even after washing with water, but none of the fabrics gave positive results for saliva after washing with detergents and alcohol.

Conclusion: From the above study it is concluded that Saliva is retained in fabrics even after washing with water. Detergents fades off the color of stain and alcohol darkens the colour of stain from study items.

This is an Open Access (OA) journal, and articles are distributed under the terms of the [Creative Commons Attribution-NonCommercial-ShareAlike 4.0 License](https://creativecommons.org/licenses/by-nc-sa/4.0/), which allows others to remix, tweak, and build upon the work non-commercially, as long as appropriate credit is given and the new creations are licensed under the identical terms.

For reprints contact: reprint@ipinnovative.com

1. Introduction

Saliva is a complex biological fluid secreted by acinar cells of the major and minor salivary glands. It is an indicator of various plasma constituents.¹ It is an important discriminating element in forensic biology, act as an indicator of salivary gland conditions, toxicological and drug monitoring.² Thiocyanate concentration in saliva can be analyzed by gas spectrometry. It is an important parameter in classifying patients as smokers and nonsmokers, determination of certain clinical conditions and in forensic drug testing.³ Extracellular nucleic acids

serve as cancer biomarkers and are detectable in a variety of biological fluids including breast milk, semen, saliva, urine, plasma as well as supernatants in cell cultures.⁴ Therefore, an improved collection method is required first to identify the invisible saliva stains on human skin and then proceed with other methods of extracting DNA to identify the suspect and exclude the innocent.⁵ Statherin and Histatin 3 are the indicators used for salivary detection.⁶ Various chemicals like enzymes and salts have also been tried out to detect dried saliva stains. Most commonly used enzymes are alkaline phosphatase, starch and amylase.⁷⁻⁹ Unfortunately, there are limitations of each test; alkaline phosphatase is not very specific as it gives a false-positive result. Starch or iodine test for amylase has been used for many years.,

* Corresponding author.

E-mail address: drneetupandey2807@gmail.com (N. Pandey).

Significant amount of saliva is deposited on the skin during biting, sucking, licking or kissing, and possibly through other behaviour.^{10,11} It can act as an important source in forensic evidence. In human's saliva is composed of 99.5% of water, 0.5 % consists of electrolytes, WBCs, epithelial cells, enzymes, lysozymes, and mucus and antimicrobial agents.¹² In forensic cases of sexual assault and child abuse, bite marks analysis is very difficult because human dentition does not always leave identifying features imprinted on the skin surface.¹³ Detection of saliva stains encountered in forensic science casework is one of the primary objectives for forensic serologist as saliva is an important source of DNA.¹⁴ Detection of saliva from human skin can be an important source for identifying an individual.¹⁵

Unfortunately dried saliva stains are invisible to the human eye which adds to difficulty in recognizing and collecting. There are so many body fluids which has been used since years. Blood is being used as main investigatory tool because of its specific color. Rapid detection and visualization of human saliva stain is one of the challenging tasks in forensic cases, such as murder, rape, sexual assault, and child abuse. A more challenging question is to detect dry saliva stains from certain difficult surfaces eg. kitchen paper towels, fabrics cloths, stainless steel spoon,(cutlery) plastic glass, paper glass because these items surfaces are considered as absorptive textured and fragile materials, rendering saliva either to be evaporated or soaked quickly. In the present study saliva stains on different fabrics viz. silk, polyester, linen, jute, cotton, georgette, sarton and other items like paper cups, chocolate wrapper aluminium, plastic bags, stainless steel spoon (cutlery) were examined using Lugol's iodine. Lugol's Iodine test was found to give Positive results for saliva stains on different fabrics and it was also found that the stains on different fabrics gives positive results for saliva even after washing with water, but none of the fabrics gave positive results for saliva after washing with detergents and alcohol. The ability of a fabric to retain stains of saliva after washing depends not only upon the chemistry and manufacturing of the fabrics but also upon the time of immersion of the fabric with water and detergent and also the method adapted to remove the stain from the fabrics. It is thus concluded that the enzyme present in saliva stick/adhere the fabrics which is generally not removed even after immersing in water and gives positive results whereas the use of detergents on different fabrics removes the stains completely and fails to give result for saliva stains and as the use of alcohol on different fabrics darkens the stains.

2. Materials and Methods

2.1. Material needed

Lugol's Iodine – (5%) This is main reagent of our study

Distilled Water- Used for washing of Saliva stained study items.

Ethanol- 90% Used for washing of Saliva stained study items.

Dropper-Used for pour Lugol's Iodine on study items.

Present study was consisted of twelve commonly found items that are expected to be present at the crime scene which are enlisted below:

Fabrics: silk, cotton, linen, jute, polyester, rayon, satin

Paper-cups/Stainless steel cutlery (spoons) Plastic bags/Aluminum foils/Chocolate wrapper

For standardization in the study, a 5cm x 5cm area were marked on each surface.

Every items put in different petri dishes. Investigator has collected her unstimulated saliva (10ml) by spitting method in a clean sterile container. All the study items has been stained with saliva and allowed to dry at room temperature by placing in numbered Petri dishes. We have made four groups on the basis of different activities.

Group A: the study items were subjected to 0.5ml of Lugol's iodine solution after a period of 12 hours, 24 hrs, 48 hrs and after 7 days.

Group A further categorized into subgroups depends upon Saliva stained study items taken in different time duration.

Group A1: All the Study items changed the colour. Its variability depends upon type of fabrics.

Group A2: Few study items were changed colour as previous, few were lighter.

Group A3: Few study items were changed colour as previous, and few were lighter.

Group A4: In this group we can easily seen even after 7 days Colour changes were seen on the salivary stained fabrics.

Group B: the study items were washed thoroughly with distilled water, allowed to dry at room temperature, and then subjected to Lugol's iodine solution.

Group C: the study items were washed thoroughly with alcohol (55%), allowed to dry at room temperature, and then

subjected to Lugol's iodine solution.

Group D: the study items were washed thoroughly with detergent (1gm of commercially available detergent [Tide] to be mixed in 500 ml of water), allowed to dry at room temperature, and then subjected to Lugol's iodine solution. All the findings were imaged using a digital camera and recorded in tables for further statistical analysis.

After washing with alcohol, saliva stains has been completely removed, but Lugol's Iodine colour became darker in fabric due to iodoform reaction occur. (When alcohol react with Lugol's Iodine give iodoform reaction). Amylase is an enzyme that breaks down starch into sugars. In amylose, sugar molecule form a structure that has small spaces in between the molecule of sugar.

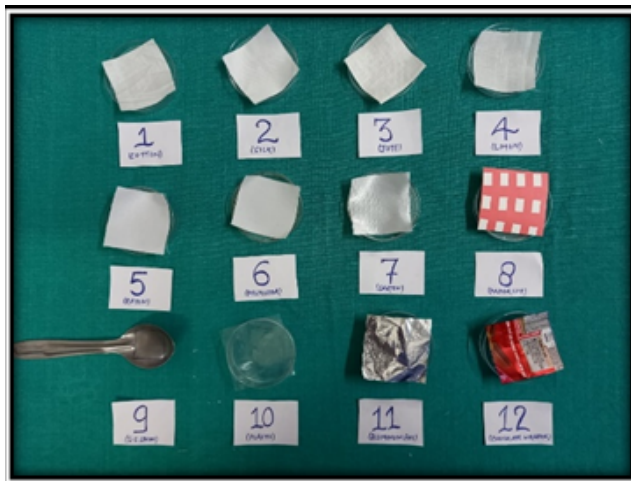


Fig. 1: Study items placed in patri dishes



Fig. 2: Study items stained with saliva



Fig. 3: Group A¹: Study items subjected to 0.5 ml of Lugol's iodine solution after a period of 12 hours and colour change is noted.



Fig. 4: Group A: Study items subjected to 0.5 ml of Lugol's iodine solution after a period of 24 hours.

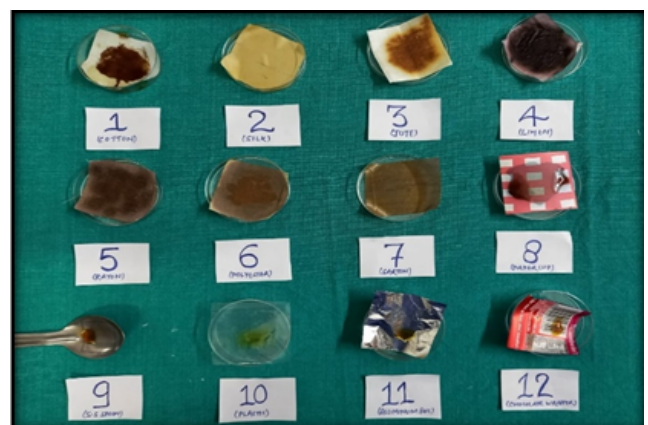
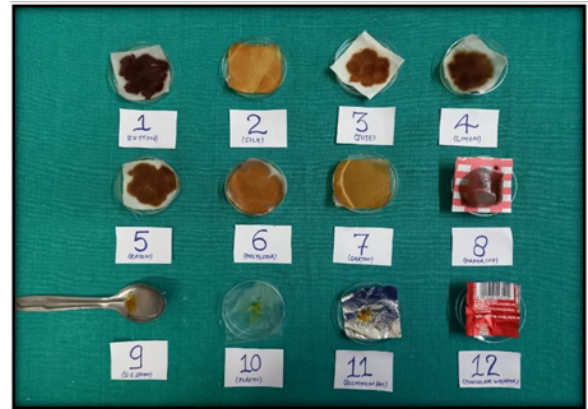


Fig. 5: Group A³: Study items subjected to 0.5 ml of Lugol's iodine solution after a period of 48 hours and colour change is noted.

Table 1: Visible color changes after application of Lugol's iodine (+: Present; -: Absent)

Study item	After 12 hours	After 24 hours	After 48 hours	After 7 days	After washing with water	After washing with detergent	After washing with alcohol
Cotton	Redish brown	Same	Same	Same	Same	Lighter	Darker
Silk	Light yellow	Lighter	Lighter	Lighter	Lighter	Lighter	Darker
Jute	Dark brown	Same	Same	Same	Same	Lighter	Darker
Linen	Brown	Same	Same	Same	Same	Lighter	Darker
Rayon	Dark Brown	Same	Same	Same	Same	Lighter	Darker
Polyester	Light brown	Lighter	Lighter	Lighter	Lighter	Lighter	Darker
Satin	Yellowish brown	Lighter	Lighter	Lighter	Lighter	Lighter	Darker
Paper-cups	Dark brown	Same	Same	Same	Same	Lighter	Darker
Stainless	Yellowish	Same	Same	Same	Same	Same	Same
Steel Spoons	Orange						
Plastic bag	Orangish Yellow	Same	Same	Same	Same	Same	Same
Aluminum Foil	Orangish yellow	Same	Same	Same	Same	Same	Same
Chocolate wrapper	Orangish yellow	Same	Same	Same	Same	Same	Same

**Fig. 6:** Group A⁴: Study items subjected to 0.5 ml of lugol's iodine solution after a period of 7 days and colour change is noted.**Fig. 8:** Group C: Study items washed thoroughly with alcohol allowed to dry at room temperature and subjected to Lugol's iodine solution and colour change is noted.**Fig. 7:** Group B: Study items washed thoroughly with distilled water allowed to dry at room temperature & then subjected to Lugol's iodine and colour change is noted.**Fig. 9:** Group D: Study items washed thoroughly with detergents (1 gm of commercially available detergents Tide to be mixed in 500 ml of water) allowed to dry at room temperature & then subjected to Lugol's iodine solution and colour change is noted.

Lugol's Iodine solution contains iodine molecules that fit tightly inside these small spaces. When the iodine molecules are inside these small spaces between bonded sugar molecules, iodine looks blue black in colour. If the sugar molecules begin to break apart and release the Iodine molecules, the indicator solution looks light brown to brown in colour. Saliva is one of the important body fluids in criminal investigation. Presence of amylase is detected by using Lugol's Iodine solution on different items. It can be detected even after washing with water. Different colour is obtained because of different webbing capacity of different clothes. Many information can be gathered in crime scene investigation through saliva examination. Saliva is generally found in sexual assault cases, or gags (cloths used to tie in mouth) in kidnapping cases or other sexually abuse cases.

3. Conclusion

Saliva is very important body fluid in criminal investigation. It is very difficult to detect because of its transparent colour. Many information can be gathered through saliva examination. From the study it is concluded that the saliva is retained in fabrics even after washing with water, due to enzymes present in saliva stick the fabrics which is generally not removed with washing with water. But after washing with detergents, It is completely removed but after washing with alcohol it becomes darkens in colour. If the criminal attempts to eliminate saliva from fabrics and different items from the crime scene. It can be still detected by using Lugol's iodine.

4. Source of Funding

None.

5. Conflict of Interest

None.

References

1. Chatterjee S. Saliva as a Forensic Tool. *J Dent Prob Solut.* 2018;5(2):26-8.
2. Aguirre A, Testa-Weintraub LA, Banderas JA, Haraszthy GG, Reddy MS. Sialochemistry: A diagnostic tool. *Crit Rev in Oral Biol and Med.* 1993;4(3-4):343-50.
3. Paul BD, Smith ML. Cyanide and thiocyanate in human saliva by gas chromatography-mass spectrometry. *J Anal Toxicol.* 2006;30(8):511-5.

4. O'driscoll L. Extracellular nucleic acids and their potential as diagnostic, prognostic and predictive biomarkers. *Anticanc Res.* 2007;27:1257-66.
5. Walsh DJ, Corey AC, Cotton RW, Forman L, Herrin GL, Word CJ. Isolation of DNA from saliva and forensic science samples containing saliva. *J Forensic Sci.* 1992;37(2):387-95.
6. Juusola J, Ballantyne J. Multiplex mRNA profiling for the identification of body fluids. *Forens Sci Int.* 2005;152(1):1-12.
7. Nelson DF, Kirk PL. The identification of saliva. *J Forensic Med.* 1963;10:14-21.
8. Willing P, Cooke WT, Nicholson GI. Globulin bound amylase. *Ann Intern Med.* 1964;60(6):1053. doi:10.7326/0003-4819-60-6-1053.
9. Auvdel MJ. Amylase level in semen and saliva stains. *J Forensic Sci.* 1986;31(2):426-31.
10. Schenkels LC, Veerman EC, Amerongen N. Biochemical composition of human saliva in relation to other mucosal fluids. *Crit Rev Oral Boil Med.* 1995;6(2):161-8.
11. Sweet D, Lorente M, Lorente JA, Valenzuela A, Villanueva E. An improved method to recover saliva from human skin: the double swab technique. *J Forensic Sci.* 1997;84:168-74.
12. Singh A, Sinha M, Sahu M, Ahirwar B. Arjun Rao Isukapatla. Examination of Tobacco Chewed saliva on different Fabrics using Starch-Iodine test: A Forensic Perspective. *Research J Pharm Tech.* 2017;10(2):4285-90.
13. Sweet D, Lorente M, Valenzuela A, Lorente JA, Alvarez JC. Increasing DNA extraction yield from saliva stains with a modified chelex method. *Forensic Sci Int.* 1996;83(3):167-77.
14. Kirk PL. New York: Interscience publishers Inc; 1953. Double swab technique. *J Forensic Sci.* 1997;83(3):168-74.
15. Nanda KD, Ranganathan K, Umadevi KM, Joshua E. A rapid and noninvasive method to detect dried saliva stains from human skin using fluorescent spectroscopy. *J Oral Maxillofac Pathol.* 2011;15(1):22-7.

Author biography

Neetu Pandey, Post Graduate Trainee

Rajendra Patil, Professor and Head

Udita Singh, Reader

Himani Singh, Senior Lecturer

Satarupa Debnath, Post graduate Trainee

Deeksha Gahlot, Post Graduate Trainee

Cite this article: Pandey N, Patil R, Singh U, Singh H, Debnath S, Gahlot D. Evaluation of lugol's iodine as a screening tool for identification of dried saliva stains. *International Dental Journal of Student's Research* 2023;11(1):5-9.